

## P11

### 無容器浮遊凝固した $\text{SiO}_2\text{-FeO}$ 非晶質の強磁性

## Magnetic properties of glass $\text{FeO-SiO}_2$ by containerless levitation method

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### 1. Introduction

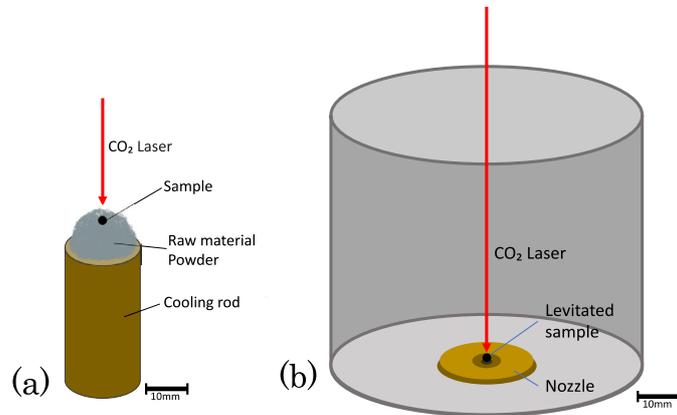
Iron containing silicates are interested in the fields of earth science, material science and steel process engineering. In materials science, magnetism by iron ions in silicates is interested, therefore the binary system of  $\text{Fe}_3\text{O}_4\text{-SiO}_2$  is investigated from the magnetics viewpoints<sup>1)</sup>. However, the binary system of  $\text{FeO-SiO}_2$  is not much investigated because magnetic properties are not expected from its equilibrium phase diagram<sup>2)</sup>. Recently, the magnetism of  $\text{FeO-CaO-SiO}_2$  system glass was reported<sup>3)</sup>. The magnetism would be  $\text{Fe}^{3+}$  ion segregation in the glass metrics. Since the  $\text{FeO-CaO-SiO}_2$  glass was easy to make glass, the initial glass samples were made by quenching of high temperature liquids using cooling plate. Then by annealing the glass samples with several conditions, magnetic properties were investigated. From this investigation, we try to make a  $\text{FeO-SiO}_2$  glass to expect the magnetism as same as  $\text{FeO-CaO-SiO}_2$  system. However,  $\text{FeO-SiO}_2$  glass has not been made by the same method of previous research. We adopted the containerless method by using aerodynamic levitation to make glass of  $\text{FeO-SiO}_2$  system in  $\text{FeO}$  rich compositions. The glass samples made under some conditions of cooling rate and the atmospheric gas showed the magnetism. In this report, we show the details of sample preparation of  $\text{FeO-SiO}_2$  glass and the results of magnetic properties measurements.

### 2. Experimental Procedure

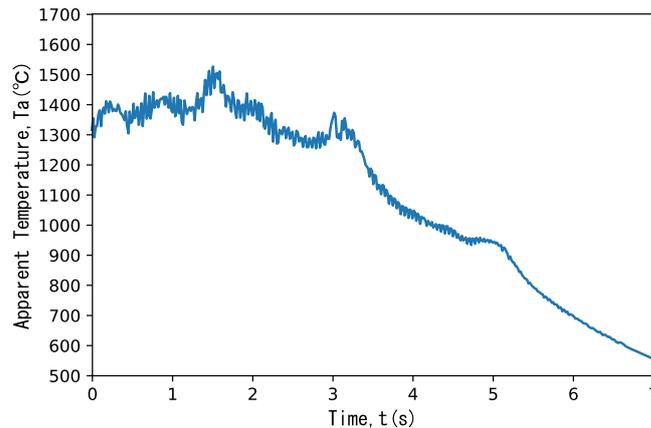
#### 2.1. Sample preparation

A master samples of  $\text{FeO:SiO}_2=40:60\text{mol}\%$  were prepared from reagent grade,  $\text{SiO}_2$ ,  $\text{FeO}$  powders (KOUJYUNDO Ltd.). Powders of  $\text{SiO}_2$  and  $\text{FeO}$  were dried at elevated temperatures to  $300^\circ\text{C}$  with 14 h in order to evaporate  $\text{H}_2\text{O}$  contained in raw powders. A weighed mixture of the oxides was placed in a Cu cooled crucible and sintered at  $1000^\circ\text{C}$  in the dried air by  $\text{CO}_2$  laser irradiations to make a shape of sample to levitate by the aerodynamic levitation (**Figure 1**). The sintered sample with a diameter of about 2mm was levitated using the aerodynamic nozzle to achieve containerless melting and solidification. Levitation conditions were in dried air and sample melting was achieved by the  $\text{CO}_2$  laser irradiations. Highest temperature for melting samples and cooling rate to solidification were variously changed. The sample temperature was monitored by

single color pyrometer with emissivity of one because the emissivity value of FeO-SiO<sub>2</sub> melt has not been obtained and reported still yet. **Figure 2** shows example of temperature profile during melting and solidification. From the result, we did not find the recalescence phenomena of crystallization from the undercooled melt with releasing latent heat of solidification. Therefore, the samples were formed glass with amorphous structure.



**Figure 1.** Schematic figure of FeO-SiO<sub>2</sub> sample preparation: (a) laser sintering method for making initial samples and (b) aerodynamic levitation method for glass samples.



**Figure 2.** Temperature profiles of cooling of aerodynamic levitated FeO:SiO<sub>2</sub> = 40:60 samples.

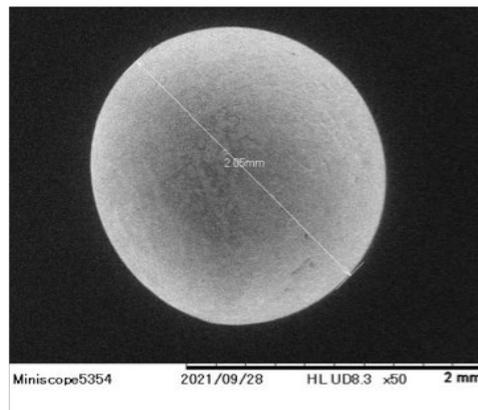
## 2.2. Measurement

Glass samples obtained by above procedures were measured the magnetic and structural properties. Magnetization measurements were carried out by the vibrating sample magnetometer (VSM) at room temperature. The solidified samples were observed by the scanning electron microscope (SEM) for observation of segregations inside samples and measured by x-ray diffraction (XRD) method for structure properties.

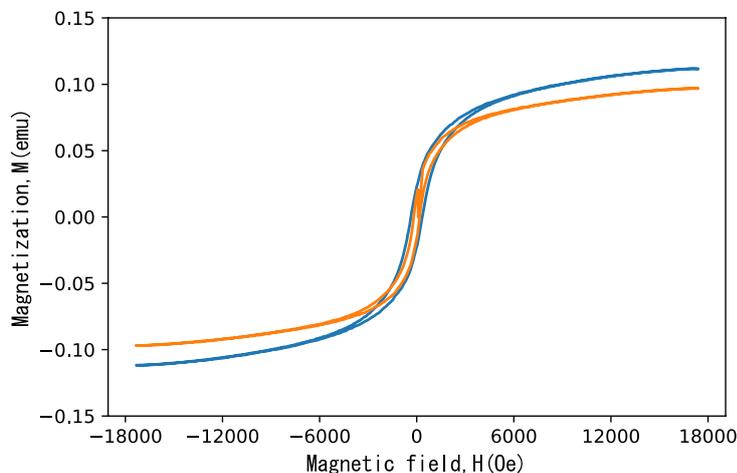
## 3. Result and discussion

**Figure 3** shows the sample shape of FeO:SiO<sub>2</sub>=40:60 mol% by optical observations. The shape is spherical and has not faced planed, therefore from the figure we identified that the sample formed glass. XRD results

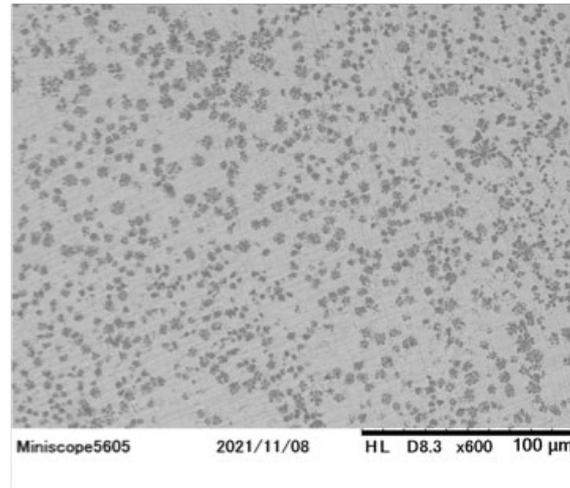
were show no peaks corresponds crystalline phase ( $\text{Fe}_2\text{SiO}_4$  and  $\text{SiO}_2(\text{tridymite})$ ) from expectation from the phase diagram for  $\text{FeO}:\text{SiO}_2=40:60$  mol% compositions. **Figure 4** shows the results of the magnetization measurements by VSM. From the results, we found the small hysteresis loops, and the other samples made by different conditions of cooling rate and atmosphere showed the small magnetization of saturations with the large hysteresis loops. **Figure5** is observation results by SEM the inside of samples shown in **Figure 4**. In **Figure 5**, some segregations are appeared in the main matrix phase. From compositional analysis by the energy dispersive x-ray analysis (EDX), these segregation parts were not corresponded to  $\text{FeO}$  and  $\text{Fe}_2\text{SiO}_4$ . If the segregation parts are caused the magnetism, the expect compositions are  $\text{Fe}_3\text{O}_4$ . However, the parts compositions were not corresponded  $\text{Fe}_3\text{O}_4$ . Therefore, it is not understood the magnetism of the samples of  $\text{FeO}:\text{SiO}_2=40:60$  mol% from these results, we tried to analyze the magnetization curve by the singularity value decomposition (SVD) method proposed by Sasayama et al.<sup>4)</sup> in to the initial part of magnetization curves shown in **Figure 4**. From the SVD results, we find that the small particles with a diameter of 10nm caused by the magnetization curve origin. The size of particle inside the main matrix phase cannot observe by SEM. From the size of magnetic particle, we attribute the Fe nanoclusters appeared in main matrix phase for the magnetism origin of the present glass samples. Therefore, for the future we need to attribute the magnetism origin by more microscopic measurement method electron spin resonance, transmission electron microscope and magnetic scattering using synchrotron radiation.



**Figure3.** SEM images of as solidified samples of  $\text{FeO}:\text{SiO}_2=40:60$ .



**Figure 4.** Magnetization of  $\text{FeO}:\text{SiO}_2=40:60$  glass samples by VSM at room temperature.



**Figure 5.** SEM images of inside of specimens shown in Figure 3 for EDX analysis.

#### 4. Conclusion

We succeed to make glass sample of FeO:SiO<sub>2</sub>=40:60 mol% by the containerless method using aerodynamic levitation with suitable colling and atmospheric conditions. The sample was observed the magnetization with hysteresis loops, but we cannot observe magnetism phase in the sample by SEM and EDX analysis. Therefore, From the present research could not be clarified the origin of magnetization of FeO:SiO<sub>2</sub>=40:60 mol% glass samples, but from the magnetization curve analysis we estimated that Fe nanoclusters in the main matrix phase segregating are origin the magnetism.

#### References

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