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Thermal Storage Project に向けた **Fe-Cu** 合金融体の表面張力測定**Surface Tension Measurement of Liquid Fe-Cu for Thermal Storage Project**殿岡和己¹, 菅原陸¹, 小島秀和², 小澤俊平¹**Kazuki TONO-OKA¹, Riku SUGAHARA¹, Hidekazu KOBATAKE², and Shumpei OZAWA¹**¹ 千葉工業大学, Department of Advanced Materials Science and Engineering, Chiba Institute of Technology,² 同志社大学, Organization for Research Initiatives & Development, Doshisha University,**1. Introduction**

Fe-Cu binary alloy shows a metastable miscibility gap in a wide composition range when the melt is undercooled below liquidus temperature. If the undercooled melt separates to form small droplet of Cu-rich phase in the matrix of Fe-rich phase at the immiscible liquid region followed by a frozen of the demixed microstructure at room temperature, it has a potential to be used as a new high-performance thermal storage material utilizing a latent heat of liquid-solid phase transformation of Cu-rich phase inside the outer supporting material of Fe-rich phase. Accurate measurement of thermophysical properties of molten Cu-Fe alloys is planned on Japanese science module for the International Space station (ISS), KIBO, as a part of ELF-Thermal Storage project¹. Accurate thermophysical properties data is crucial to control the microstructure of Fe-Cu alloys.

In the present study, we investigated the effect of changes in the sample composition due to evaporation at high temperature on the surface tension of molten Fe-Cu alloy by using the electromagnetic levitation (EML) on the ground as a preliminary study for the ELF-Thermal Storage project¹ at the KIBO, because one of the components in the alloy may preferentially lost during the measurement. Furthermore, the influence of oxygen partial pressure, P_{O_2} , on the surface tension was confirmed because oxygen is one of the strongest surfactants for molten metallic melt.

2. Experimental Procedure

A cube of high purity iron with a nominal composition of 99.99 mass% was electromagnetically levitated with a cube of copper with a nominal composition of 99.99 mass% simultaneously and then uniformly melted under a flowing of Ar-He and Ar-He-0.5 vol.%H₂ gases. After the indicated values of sample temperature and oxygen partial pressure of the atmospheric gas, measured by pyrometer and zirconia oxygen sensor, became constant, the oscillation behavior of the droplet was monitored from above at 500 fps for 16 s using a high-speed video camera. The frequencies of the surface oscillations and those of the center of gravity were analyzed

from the HSV images using fast Fourier transformation. The surface tension of the droplet was calculated from these frequencies using the Rayleigh²⁾ equation and the Cummings and Blackburn³⁾ calibration. The sample composition was analyzed by inductively coupled plasma (ICP).

3. Results and discussion

Figure 1 shows the surface tension of molten Fe–Cu alloys measured by EML. Even though an initial copper composition of the sample was 10 at% (▲), it decreased with elevating the temperature due to the evaporation under Ar–He–0.5 vol.%H₂ gas. As a result, the surface tension of molten sample shows almost the same value regardless of temperature, as shown by a green line. When the copper composition after the measurement was controlled to be 10 at% even at high temperature by adjusting the initial composition and maintained period (●), the surface tension was decreased with elevating temperature as shown by a blue solid line.

Although the P_{O_2} was increased to 10^{-2} Pa under Ar–He gas (◆), the surface tension of molten sample showed almost the same as that measured at $P_{O_2} < 10^{-10}$ Pa under Ar–He–0.5 vol.%H₂ gas, when the copper content in the sample was 10at% after the measurement. It was reported that the surface tension of molten copper is not affected by oxygen adsorption under Ar–He gas at P_{O_2} of 10^{-2} Pa⁴⁾. The surface tension of molten Fe–10at%Cu alloy was clearly lower than the averaged surface tension calculated from the surface tension of liquid iron and copper shown by a blue dotted line, suggesting a surface segregation of copper. Therefore, surface tension of molten Fe–Cu alloy would not be affected by oxygen adsorption under Ar–He gas at P_{O_2} of 10^{-2} Pa.

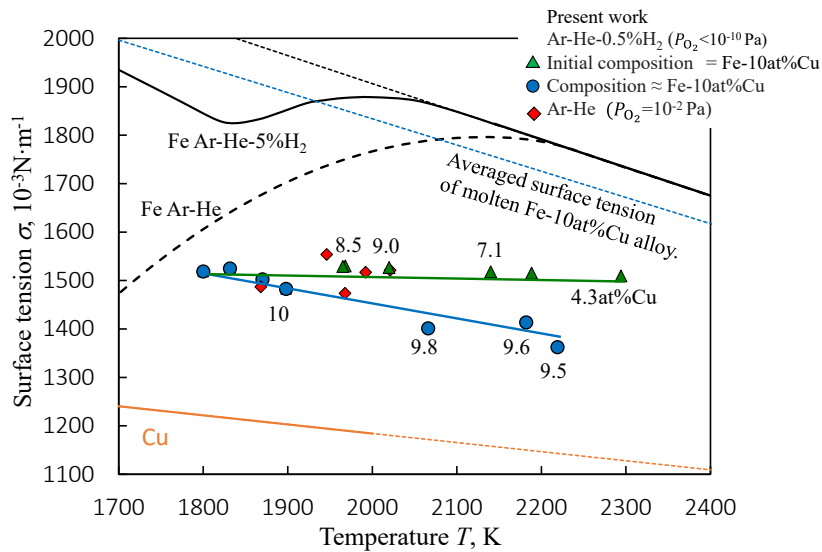


Figure 1. Surface tension of molten Fe–Cu alloys along with literature data of surface tension of liquid iron⁵⁾ and copper⁴⁾. The numbers near the plot indicate a copper content in the sample after measurement.

4. Summary

The surface tension of Fe–Cu alloy was measured by oscillating droplet method using the EML. The surface tension of molten Fe–10at%Cu decreased with elevating temperature at $P_{O_2} < 10^{-10}$ Pa under Ar–He–0.5vol.%H₂ gas. However, the decrease in the copper content due to the evaporation at high temperature increased the surface tension. Even though the P_{O_2} was increased to 10^{-2} Pa under Ar–He gas, it did not affect the surface tension of the molten when the copper content was 10at% after the measurement.

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